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Synthesis and Biological Evaluation of Cyclophostin: A 5',6"-Tethered Analog of Adenophostin A

Martin de Kort,^a Anouk D. Regenbogen,^a Herman S. Overkleeft,^a R. A. John Challiss,^b Yoriko Iwata,^c Shuichi Miyamoto,^c Gijs A. van der Marel^a and Jacques H. van Boom^{a,*}

a Leiden Institute of Chemistry, Gorlaeus Laboratories, Leiden University, P.O. Box 9502, 2300 RA Leiden, The Netherlands

b Department of Cell Physiology & Pharmacology, Maurice Shock Medical Sciences Building, University of Leicester,

University Road, Leicester, LE1 9HN, UK

c Exploratory Chemistry Research Laboratories, Sankyo Co., Ltd., 1-2-58 Hiromachi, Shinagawa-ku, Tokyo 140-8710, Japan

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Abstract—The synthesis, conformational analysis and biological evaluation of $5'$ -6"-tethered adenophostin A, so-called cyclophostin 14, and its de-adeninylated analog 15 are described. They are prepared via ring-closing metathesis of diolefin 28, consecutive coupling of the central building block 33 to 6-N-benzoyladenine or propargyl alcohol, respectively, followed by phosphorylation and deprotection. NMR spectroscopy and a molecular dynamics simulation indicated that the $5'$ -6"-tether induces a conformational change from $2'$ -endo/syn in 1 to $3'$ -endo/anti in 14. The unexpected small loss of Ca²⁺-releasing potency of cyclophostin 14, which is reflected by the low EC₅₀/IC₅₀ ratio in comparison with cycloribophostin 15, suggests that the interaction of the adenine with IP₃R plays a decisive role in determining the high activity of adenophostin A (1). \oslash 2000 Elsevier Science Ltd. All rights reserved.

Introduction

The gluconucleosides adenophostin A and B (1 and 2) exhibit \sim 10-100 times higher Ca²⁺-mobilizing potencies in comparison with the natural second messenger $D-m\nu\ddot{o}$ inositol 1,4,5-trisphosphate $(\text{IP}_3, \text{4})$.¹ Despite many synthetic efforts in determining the minimal structural requirements for the high activities of 1 and 2 , no simplified analogs displaying a higher activity than IP₃ (4) have been reported. The observed $1c \sim 1000$ -fold reduced binding affinity of the $2'$ -dephosphorylated derivative 3 and the IP_3 -like potencies found for the de-adeninylated analogs $5-7^2$ show that the extremely high potency of adenophostin A (1) is mainly governed by the adenine and the $2'$ -phosphate moieties. Additional studies have revealed that the $Ca²⁺$ -releasing activity of conformationally flexible analogs of 1 is significantly reduced. For instance, the hydroxyethyl glucosides 9^3 and 10^4 exhibit ~10-fold diminished potencies relative to 5, whereas acyclophostin $11⁵$ is a pHdependent partial agonist. It has been proposed that the tripodal arrangement of the phosphates in adenophostin A (1) may stabilize a long-range interaction with the IP₃ receptor (IP_3R) .⁶ The latter is endorsed by molecular modeling^{7,3b} studies, which showed that the $2'$ -phosphate in 1 occupies a more remote position from the $trans-3ⁿ,4ⁿ$ -bisphosphate

than the corresponding 1-phosphate in IP_3 (4). We recently reported⁸ that the \sim 20-fold decreased activities of the rigid analogs spirophostin (3R)-12 and (3S)-13 relative to IP₃ (4) can be ascribed to an improper spatial orientation of the non-vicinal 3-phosphates. The aforementioned observations imply that a cooperative effect of the adenine moiety and the 2'-phosphate may be responsible for the enhanced activity of adenophostin A (1). It was envisaged that a non-deadeninylated analog in which the conformational freedom of the isolated 2'-phosphate function is restricted would be of value in probing the possible existence of the proposed cooperative effect of the adenine.

In this paper, we describe the synthesis of the conformationally restricted adenophostin 14, so-called cyclophostin, and its de-adeninylated analog 15. In addition, the biological activity of both analogs in terms of conformational behavior is assessed by NMR spectroscopy and molecular modeling (Fig. 1).

Results and Discussion

Comparison of the biological activities of furanophostin 7^9 and ribophostin $6²$ as well as those of xylose-derivative 10 and glucopyranose 9^3 , shows that HO-5^{\prime} and HO-6^{$\prime\prime$} do not contribute to the activity of 1. Moreover, a recent molecular modeling study⁷ of adenophostin A (1) revealed that both hydroxyls are in close proximity, inferring that the rotational freedom of 1 can be limited by anchoring either

Keywords: adenophostin A; gluconucleoside; ring-closing metathesis; calcium release; molecular modeling.

^{*} Corresponding author. Tel.: $+31-71-527-4375$; fax: $+31-71-527-4307$; e-mail: j.boom@chem.leidenuniv.nl

Figure 1. Structures of adenophostin A (1) and B (2), IP₃ (4), ribophostins 5–8, acyclophostin 11, spirophostins 12 and 13, and cyclophostins 14 and 15.

HO-5^{\prime} to HO-6^{$\prime\prime$} or CH₂-5^{\prime} to CH₂-6^{$\prime\prime$} with an appropriate alkyl spacer. To this end, attention was focused on the synthesis of disaccharides $26-28$ (see Scheme 1) which, after ring-closing metathesis¹⁰ (RCM), would lead to the respective eight-, eleven- and fourteen-membered ring derivatives 29-31.

The synthesis of key dienes 26 and 27 commenced with N -iodosuccinimide (NIS)-mediated¹¹ coupling of known olefin 16^{12} with thioglucopyranoside 18, prepared by tritylation and subsequent benzylation of known 17,¹⁶ to afford the α -linked disaccharide 22 in 86% yield. Detritylation of 22 under the influence of p -toluenesulfonic acid gave disaccharide 23, which was readily converted to diene 26 by Swern oxidation of the primary hydroxyl group and subsequent Wittig olefination. Alternatively, allylation of 23 gave the $6'$ -O-allyl derivative 27. Having the dienes 26 and 27 in hand, the formation of the eight- and eleven-membered rings 29 and 30 by RCM using the Grubbs' catalyst¹⁴ $(PCy_3)_2Cl_2Ru=CHPh$ was undertaken. It was established that the bis-vinyl substrate 26 failed to give the eightmembered ring 29 under a variety of reaction conditions. On the other hand, RCM of the $6'$ -O-allyl-4-vinyl derivative 27 led to the exclusive formation of the unwanted $6'$ -tethered dimer in 69% yield. The outcome of these experiments may be explained by the steric congestion of the vinyl functions¹⁵ and an indomitable ring strain.¹⁶ In order to overcome the problems encountered in the above cyclizations, the sterically less demanding $5,6'$ -di- O -allyl derivative 28 was used for the construction of the fourteen-membered macrocycle 31 (see Scheme 1). Thus, coupling of 5-O-trityl-1,2-isopropylidene- α -D-ribofuranoside²⁶ (20) with donor 18 gave the α -linked dimer 24 $(R=Tr)$. However, ensuing detritylation of 24 $(R=Tr)$ under the influence of 1% p-TsOH in CH₂Cl₂/MeOH (1/1, v/v) was accompanied by concomitant cleavage of the glucosidic bond. The latter disadvantage could be circumvented as follows. Condensation of partially protected

acceptor 21^{17} with donor 19, prepared by silylation and benzylation of 17, gave dimer 24 which, in turn, was readily transformed into diene 28 by desilylation $(\rightarrow 25)$ and allylation in 72% yield over the last three steps. Gratifyingly, subjection of 28 to RCM proceeded uneventfully to give 31 as a mixture of E/Z isomers. Selective hydrogenation (Scheme 2) of the olefin bond in 31 with H_2/PtO_2 gave homogeneous 32, the structure of which was fully ascertained by NMR spectroscopy and mass spectrometry. Acid-mediated deprotection¹³ of 32 and subsequent acetylation afforded key building block 33. Vorbrüggen-type condensation of 33 with silylated N^6 -benzoyladenine in the presence of trimethylsilyl trifluoromethanesulfonate (TMSOTf) gave the tethered adenosine derivative 34 as a single isomer. The introduction of the three phosphate groups was performed by the following three-step sequence of reactions.¹⁷ Selective deacetylation of 34 by short treatment with potassium t-butoxide in MeOH, followed by 1H-tetrazole-assisted phosphitylation of the alcohol functions with N,N-diisopropyl-bis-[2-(methylsulfonyl) ethyl] (MSE) phosphoramidite, 18 and in situ oxidation of the intermediate phosphite triesters with t-butylhydroperoxide, gave trisphosphate 35. Deprotection of the latter was affected in two consecutive steps involving removal of the base-labile groups with Tesser's base, 19 and hydrogenolysis of the 2'-O-benzyl ether to afford, after purification by HW-40 gel filtration and Dowex-ion exchange chromatography, homogeneous cyclophostin 14 (Na⁺-salt). The deadeninylated analog of 14, i.e. cycloribophostin 15, was prepared via deacetylation of propargyl glycoside 36, obtained by glycosylation of propargyl alcohol with tetraacetate 33, and phosphitylation of the intermediate triol with N , N -diisopropyl-dibenzyl phosphoramidite,²⁰ to furnish trisphosphate 37. Debenzylation of 37 by hydrogenolysis, followed by purification as described earlier, gave homogeneous target compound 15. The identity of 14 and 15 was fully ascertained by ${}^{1}H$, ${}^{13}C$ and ${}^{31}P$ NMR spectroscopy, as well as by high resolution mass spectrometry.

Scheme 1. Reagents and conditions: (i) a. TrCl, pyridine, 16 h, 85%; b. NaH, BnBr, DMF, 2 h, 93%. (ii) a. TBDPSCl, pyridine, 16 h, 78%; b. NaH, BnBr, DMF, 2 h, 91%. (iii) NIS, TfOH, Ms 4 Å, Et₂O or toluene/1,4-dioxane, 1/3, v/v, 1 h, 22: 86%, 24: 85%. (iv) 1% p-TsOH, CH₂Cl₂/MeOH, 1/1, v/v, 4 h, 84%. (v) a. (C(O)Cl)₂, Et₃N, DMSO, CH₂Cl₂, -60° C, 2 h; b. MePPh₃Br, *n*-BuLi, THF, 0°C, 2 h, 71%, 2 steps. (*vi*) NaH, allylbromide, DMF, 3 h, **27**: 70%; **28**: 89%. (vii) 1 M TBAF, THF, 50°C, 16 h, 95%. (viii) 5 mol% RuCl₂(PCy₃)₂CHPh, 0.025 mM in toluene, 16 h, 29: 0%, 30: 0% (69% of 6'-O-allyl dimer formed), 31: 75%.

Scheme 2. Reagents and conditions: (i) PtO₂, H₂, 2 h, quant. (ii) a. HOAc/H₂O/(HOCH₂)₂, 14/6/3, v/v/v, reflux, 90 min.; b. Ac₂O/pyridine, 4 h, 92%, 2 steps. (iii) a. HMDS, pyridine, 6-N-benzoyladenine, 7 h; b. TMSOTf, 1,2-dichloroethane, reflux, 16 h, 82%. (iv) propargyl alcohol, TMSOTf, (CH₂Cl)₂, 6 h, 82%. (v) a. t -BuOK (1 M in MeOH), 1 min; b. (MSEO)₂PN(i -Pr)₂, 1H-tetrazole, 30 min; c. t -BuOOH, 30 min. (vi) a. NaOH (4 M)/1,4-dioxane/MeOH, 1/14/5, v/v/v, 16 h; b) H2, Pd-black, H2O, 16 h, 56% (based on 34). (vii) a. NaOMe, MeOH, 1 h; b. (BnO)2PN(i-Pr)2, 1H-tetrazole, 30 min; c. t-BuOOH 30 min, 60%. (viii) H2, 10% Pd/C, 1,4-dioxane/i-PrOH/H2O (4/2/1, v/v/v), 16 h, 84%.

^a See Ref. 7.

 b Calculated with the conformational analysis program Pseurot v6.3. 30

^c Measured at 600 MHz in 50 mM phosphate buffer at p^2H 6.8.
^d Measured at 300 MHz in D₂O.

^c Measured at 600 MHz in 50 mM phosphate buffer at p²H 6.8.
^d Measured at 300 MHz in D₂O.
^e Based on the assumption that $10 \times \mathcal{Z} J_{1',2'}$ represents the percentage of 2^{*'}-endo* (S-type) conformer.^{21,22}</sup>

Conformational Analysis

In the first instance, the ${}^{1}H$ NMR spectroscopic data of cyclophostin 14 and its de-adeninylated analog 15 were compared with those of adenophostin A (1) .⁷ It turned out that the glucose moiety in 14 and 15 adopts, based on the coupling constants of the glucose ring protons (see Experimental section), a 4C_1 conformation as in 1. In contrast, the coupling constants of the ribose protons (see Table 1) clearly show^{21,22} that the C-3'-endo conformation in both 14 and 15 prevails over the C-2'-endo conformation of adenophostin A (1). The same holds, gauged by the absence of a vicinal coupling between H-1' and H-2' $(J_{1',2'} \sim 0)$, for ribophostin 8.^{2c,13}

As can be seen in Fig. 2, the NOEs between H-8 of the adenine and $H-1'$, $H-2'$ and $H-3'$ of the ribose ring clearly show that the nucleobase in cyclophostin 14 adopts an *anti* conformation. Moreover, the earlier assigned preference of 14 in adopting the C-3'-endo conformation is endorsed by the absence of a NOE signal between $H-1'$ and $H-4'$.

The effect of the $5^{\prime}, 6^{\prime\prime}$ -tether on the position of the 2^{\prime} -phos-

phate $(P-2)$ and the conformational rigidity was investigated by a molecular dynamics simulation of cyclophostin 14. The data of the global minimum conformation, presented in Fig. 4A and Table 2 (entry 3), clearly indicate that the distances between the $trans-3''$, $4''$ -bisphosphate and P-2^{\prime} do not deviate substantially from those observed⁷ in adenophostin A (1). The different values of τ_c and χ (see Table 2 and Fig. 3) of the global minimum 3'-endolanti conformation of cyclophostin 14 are in accord with its NMR-spectroscopic data (see Table 1). Superimposition of cyclophostin 14 and adenophostin A (1), as illustrated in Fig. 4A, showed that the distance between the individual $2'$ -phosphates is 1.4 Å. Furthermore, it can be seen that the triangular arrangement of the phosphates in 14 is slightly smaller than the phosphorous triad in 1 and that the distances between C-5^{\prime} and C-6^{$\prime\prime$} are the same (i.e. 4.4 Å, see entries 2 and 3 in Table 2). It can also be derived from the superimposition of the four lowest energy conformers of 14 (see Fig. 4B) that the conformation of the tethered disaccharide is constrained, whereas the adenine nucleobase adopts several preferred positions in the anti conformation. Apart from this, the presence of three conformers of cyclophostin 14 having a 2'-endo conformation similar to

Figure 2. Part of the 600 MHz NOESY spectrum of cyclophostin 14 (10 mM) in 50 mM phosphate buffer in D₂O at p^2H 6.8.

Table 2. Conformational data for adenophostin A (1) , IP₃ (4) and cyclophostin $(14).^{29}$

^a Modeling of IP₃ and adenophostin A: see Ref. 7.
^b Conformational energies relative to most stable conformer in entry 3 (kJ mol⁻¹).
^c Numbering for IP₃.

Figure 3. Torsion angles τ and χ assigned in molecular modeling of adenophostin A (1, R=H) and cyclophostin 14 [R,R=(CH₂)₄].

adenophostin $A(1)$ was established (entries 4–6). However, the latter conformers are less stable (i.e. \sim 17 kJ mol⁻¹) than the energy-minimized 3'-endo-puckering structure (entry 3).

In summary, based on the molecular modeling results it may be concluded that the $3'-O$ - α -glucosyl ribose moiety in cyclophostin 14 closely resembles the geometry of the glucosidic bond in adenophostin $A(1)$ and that the distance between the $C-5'$ and $C-6''$ positions is preserved. On the other hand, the adenosine moiety undergoes a conformational change from 2'-endo/syn in 1 to 3'-endo/anti in 14.

Biological Evaluation

The IC_{50} values (see Table 3) of cyclophostin 14 and cycloribophostin 15, as well as those of adenophostin A (1), IP_3 (4) and ribophostin $8^{2c,13}$ were determined in 3 H-IP₃ displacement binding experiments using bovine adrenal cortex membranes.²³ The relative displacing potencies of adenophostin A (1) and IP₃ (4) (entries 1 and 2) are consistent with earlier reported data while those of cyclophostin 14 (entry 3) and ribophostin 8 (entry 4) are similar to IP_3 . The deadeninylated analogs ribophostin 8 and cycloribophostin 15 (entry 5) both exhibit a \sim 20-fold decreased binding affinity to IP₃R in comparison with adenophostin A $(1, \text{entry } 2)$ and

cyclophostin 14 (entry 3), respectively. Moreover, the $5', 6''$ tethered ligands 14 and 15 display a consistent \sim 15-fold diminished binding affinity in comparison with adenophostin A (1) and ribophostin 8.

A functional response to cyclophostins 14 and 15 was studied by measuring the ${}^{45}\text{Ca}^{2+}$ -release from intracellular stores upon binding to IP3Rs in permeabilized SH-SY5Y neuroblastoma cells²⁴ in comparison with the activities of adenophostin A (1) , IP₃ (4) and ribophostin **8** (see Table 4). Perusal of Table 4 reveals that the relative potencies and slope (h) of the concentration-response curves of all analogs are in agreement with the binding data presented in Table 3. Interestingly, the Ca^{2+} -mobilizing potency of cyclophostin 14 (entry 3) is now only \sim 5-fold decreased in comparison with adenophostin $A(1)$, whereas the tethered ribophostin derivative 15 is \sim 13-fold less effective than the corresponding ribophostin 8.

The respective binding and Ca^{2+} -release data in Tables 3 and 4 were compared by calculating the EC_{50}/IC_{50} ratios (see Table 4), which give an indication of the efficacy of each of the ligands. The relatively high value for IP₃ (4) (entry 1) can perhaps be ascribed, despite the short (30 s) incubation period, to partial metabolism of IP_3 in the ${}^{45}Ca^{2+}$ -assay. The metabolically stable adenophostin A (1, entry 2) and its de-adeninylated analog 8 (entry 4) have similar EC_{50}/IC_{50} ratios. Despite the substantial loss of biological activity, the efficacy of the tethered cycloribophostin 15 (entry 5) is entirely predictable. In contrast, the EC₅₀/IC₅₀ ratio of cyclophostin 14 (entry 3) is \sim 3-fold reduced, illustrating that the Ca^{2+} -releasing potency of 14 is less affected than the binding affinity for IP_3R .

In summary, the binding affinity of cyclophostin 14 and cycloribophostin 15 is in both cases \sim 15-fold diminished in comparison with adenophostin A (1) and ribophostin 8, respectively. However, the Ca^{2+} -releasing activity of 14 is only \sim 5-fold lower than that observed for 1 (still \sim 5-fold higher than IP₃), which is in contrast to the \sim 13-fold reduced potency of 15, relative to 8.

Figure 4. Graphical representation of the low-energy conformers with superimposition of the six-membered rings: A. global energy minimized conformations of cyclophostin 14 (blue) and adenophostin A (1); B. superposition of the four lowest-energy conformers of cyclophostin 14.

Table 3. ³H-IP₃-displacement/binding IC₅₀ values of cyclophostins **14** and **15** in comparison with IP₃ (4), adenophostin A (1) and ribophostin **8** (values are shown as \pm s.e. mean for the concentration which causes 50% of specific 3 H-IP₃ displacement (IC₅₀), with h as the slope of the concentration-response curve, for n experiments)

Entry	Compound	$-\log$ IC ₅₀	IC_{50} value (nM)	Ratio		n	
	$IP_3(4)$	8.13 ± 0.05	7.4	18	0.99 ± 0.01		
2	Adenophostin $A(1)$	9.39 ± 0.07	0.41		1.51 ± 0.13		
3	Cyclophostin (14)	8.23 ± 0.05	5.9	14	1.20 ± 0.05		
4	Ribophostin (8)	8.08 ± 0.09	8.3	20	0.79 ± 0.03		
5	Cycloribophostin (15)	6.94 ± 0.04	114	277	0.80 ± 0.02		

Table 4. $^{45}Ca^{2+}$ -release EC₅₀ values of cyclophostins 14 and 15 in comparison with IP₃ (4), adenophostin A (1) and ribophostin 8 (values are shown as \pm s.e. mean for the concentration which causes 50% of maximal ⁴⁵Ca²⁺ release (EC₅₀), with h as the slope of the concentration-response curve, the % release is relative to ionomycin-induced Ca^{2+} release, for *n* experiments)

Conclusion

The synthesis of the fourteen-membered macrocycle cyclophostin 14, as well as its de-adeninylated analog 15, has been accomplished via ring-closing metathesis of carbohydrate diene 28. NMR spectroscopy and molecular modeling studies of these conformationally restricted analogs of adenophostin A (1) revealed that the $5^{\prime}, 6^{\prime\prime}$ -O-n-butyl tether in 14 and 15 induces a C-3'-endolanti conformation, rather than a C-2'-endo/syn conformation as in 1. Although the mode of binding is at present not fully understood, the \sim 5-fold higher Ca²⁺-releasing potency relative to IP₃ (4) indicates that cyclophostin 14, despite the presence of a conformational restraint, closely resembles the bioactive conformation of adenophostin A (1). Moreover, the unexpected small loss of Ca^{2+} -mobilizing potency of cyclophostin 14, which is reflected by the low EC_{50}/IC_{50} ratio in comparison with cycloribophostin 15, shows that the loss of activity caused by the $5^{\prime}, 6^{\prime\prime}$ -tether can still be compensated by the adenine nucleobase. The latter implies that the adenine may have a more decisive influence on the enhanced activity of adenophostin A (1) than the orientation of the 2'-phosphate.²⁵ The cooperative effect of the adenine and the 2'-phosphate may be studied in more detail by varying the length of the 4C-tether in 14 or limiting the conformational freedom of the nucleobase by replacing the adenine with 8-bromoadenine.

Experimental

General methods and materials

 CH_2Cl_2 and toluene were dried by distillation from P_2O_5 (5 g L^{-1}) . Et₃N and pyridine were refluxed for 2 h in the presence of CaH_2 (5 g L⁻¹) and subsequently distilled. 1,2-dichloroethane (p.a. Rathburn), 1,4-dioxane (p.a. Baker), i-propanol (p.a. Baker), DMF (p.a. Baker), DMSO (p.a. Baker), THF (Acros) were stored over molecular sieves 4 Å . CH₃CN (p.a. Rathburn) and MeOH (HPLC-grade, Rathburn) were stored over molecular sieves 3 Å. Acetic acid (p.a. Baker) and acetic anhydride (p.a. Baker) were

used as received. Allyl bromide, butane-2,3-dione, camphorsulfonic acid, ethylene glycol, trifluoromethane-sulfonic acid, oxalyl chloride, triphenylmethyl chloride, sodium hydride, t-butyldiphenylsilyl chloride, tetrabutylammonium fluoride and propargyl alcohol (Acros), $Dowex^{\circledR}$ 50WX4, methyltriphenylphosphonium bromide and di-t-butyl peroxide (Fluka), N-iodosuccinimide, trimethylsilyl trifluoromethanesulfonate and 10% palladium on charcoal were purchased from Aldrich. Benzyl bromide, imidazole and p-toluenesulfonic acid (Merck) were used as received. N,N-Diisopropyl-bis-[2-(methylsulfonyl)ethyl] phosphoramidite¹⁸ and N,N-diisopropyl-dibenzyl phosphoramidite²⁰ were prepared as described. All experiments were performed under anhydrous conditions at room temperature unless stated otherwise. Reactions were followed by TLC analysis conducted at Schleicher and Schüll DC Fertigfolien (F 1500 LS 254). Compounds were visualized by UV light and by spraying with 20% sulfuric acid in MeOH followed by charring at 140° C. Column chromatography was performed on silica gel 60 , $0.063-0.200$ mm (Baker). NMR spectra were recorded with a Bruker WM-200 (1 H and 13 C at 200 and 50.1 MHz, respectively) and a Bruker WM-300 spectrometer (${}^{1}H$ at 300 MHz). ${}^{1}H$ - and ${}^{13}C$ -chemical shifts are given in ppm (δ) relative to tetramethylsilane as an internal standard. Mass spectra were recorded on a Finnigan MAT TSQ-70 or PE-SIEX API 165 mass spectrometer equipped with an electrospray inonization (ES) interface. HRMS (ES) spectra were measured with a Finnigan MAT 900 double focusing mass spectrometer equipped with an ES interface. The samples of the target compounds 14 and 15 were prepared in a mixture of isopropanol/H2O (80/20, v/ v) containing 1.0×10^{-4} M NaOAc, the clusters of which were used as internal standards in the negative ion detection mode.

 $(2'S, 3'S)$ Ethyl 2-O-benzyl-3,4-di-O- $(2', 3'$ -dimethoxybutane-2',3'-diyl)-1-thio-6-*O*-trityl-β-D-glucopyranoside (18). Trityl chloride (5.19 g, 18.5 mmol) was added to a solution of 17^{13} (5.71 g, 16.8 mmol) in pyridine (25 mL) and the mixture was stirred overnight at 50° C. The reaction mixture was quenched with MeOH (1.0 mL) and concentrated. The residue was dissolved in EtOAc (100 mL) and washed with

sat. aq. solution of NaHCO₃ (2×25 mL) and H₂O (25 mL). The organic layer was dried over $MgSO₄$, filtered and concentrated. Traces of pyridine were removed by coevaporation with toluene $(2\times10 \text{ mL})$. The residue was applied onto a column of silica gel (Et₂O/light petroleum/Et₃N, 66/33/1, v/v/v) to yield the tritylated product as a white solid (6.69 g, 11.6 mmol). Mp 74°C. R_f 0.68 (Et₂O/light petroleum, 2/1, v/v). ¹³C{¹H} NMR (CDCl₃): δ 143.5 (Cq Ph), 128.4-125.0 (CH arom), 99.3, 99.0 (2×Cq BDA), 85.8 (C-1), 77.0, 73.7, 69.5, 65.2 (C-2, C-3, C-4, C-5), 61.6 $(C-6)$, 47.6 (2×OCH₃ BDA), 23.5 (CH₂ SEt), 17.4, 17.2 $(2\times CH_3$ BDA), 15.3 (CH₃ SEt). Sodium hydride (60% wt, 0.69 g, 17.3 mmol) was added to a cooled $(0^{\circ}C)$ solution of tritylated 17 (6.69 g, 11.6 mmol) in DMF (60 mL), followed by the addition of benzyl bromide (1.50 mL, 12.7 mmol). When TLC analysis showed complete conversion of starting material, excess sodium hydride was destroyed with MeOH $(0.5$ mL) and the reaction mixture was diluted with $Et₂O$ (50 mL) . H₂O (15 mL) was added and the DMF/H₂O layer was extracted with $Et₂O$. The organic layer was rinsed with $H₂O$ (2×15 mL), dried over MgSO₄, filtered and concentrated in vacuo. Compound 18 was obtained as a white solid by crystallization from $Et_2O/light$ petroleum. Yield 7.42 g (11.0 mmol, 65% over two steps). Mp 69 $^{\circ}$ C. $R_{\rm f}$ 0.85 (Et₂O/light petroleum, 1/1, v/v). ¹H NMR (300 MHz, H-H-COSY, CDCl₃): δ 7.52–7.18 (m, 20H, H arom), 4.87 $(s, 2H, CH₂ Bn), 4.54$ (d, 1H, H-1, $J_{1,2}=9.5$ Hz), 3.95 (t, 1H, H-2, $J_{2,3}$ =9.5 Hz), 3.84 (t, 1H, H-3, $J_{3,4}$ =8.8 Hz), 3.65–3.41 (m, 3H, H-4, H-5, H-6a), 3.28 (s, 3H, OCH₃ BDA), 3.13- 3.08 (m, 1H, H-6b), 3.06 (s, 3H, OCH₃ BDA), $2.99-2.70$ (m, 2H, CH₂ SEt), 1.42-1.27 (m, 3H, CH₃ SEt), 1.36, 1.16 (2 \times s, 6H, 2 \times CH₃ BDA); ¹³C{¹H} NMR (CDCl₃): δ 143.7 $(Cq Tr)$, 138.0 $(Cq Bn)$, 128.4–126.5 (CH arom), 99.2, 99.0 (2£Cq BDA), 85.9 (Cq Tr), 84.3 (C-1), 78.1, 76.7, 74.6, 65.3 (C-2, C-3, C-4, C-5), 75.0 (CH₂ Bn), 61.4 (C-6), 47.8, 47.4 (2×OCH₃ BDA), 23.8 (CH₂ SEt), 17.5, 17.1 $(2\times CH_3$ BDA), 15.0 (CH₃ SEt). C₄₀H₄₆O₇S (671): Calcd C 71.62, H 6.91; found C 71.73, H 6.96.

 $(2'S, 3'S)$ Ethyl 2-O-benzyl-6-O-t-butyldiphenylsilyl-3,4di-O-(2',3'-dimethoxybutane-2',3'-diyl)-1-thio-β-D-glucopyranoside (19). t-Butyldiphenylsilyl chloride (2.30 mL, 8.87 mmol) was added to a solution of 17^{13} (2.51 g, 7.39 mmol) in pyridine (15 mL) and the mixture was stirred for 2 h. MeOH (0.4 mL) was added and the mixture was concentrated. The residue was dissolved in EtOAc (50 mL) and washed with sat. aq. solution of NaHCO₃ (15 mL) and H_2O (15 mL) . The organic layer was dried over MgSO₄, filtered and concentrated. Traces of pyridine were removed by coevaporation with dry toluene (2×10 mL). The crude product (R_f 0.67, Et₂O) was benzylated as described for 18. Compound 19 was purified by column chromatography (EtOAc/light petroleum, $1/10 \rightarrow 1/$ 1, v/v) and obtained as a colorless oil. Yield 4.40 g (6.58 mmol, 89% over two steps). R_f 0.79 (Et₂O/light petroleum, $1/1$, v/v). ¹H NMR (300 MHz, H-H-COSY, CDCl₃): δ 7.75–7.66 (m, 4H, H arom), 7.47–7.21 (m, 11H, H arom), 4.87 (AB, 2H, CH₂ Bn, $J=-12.3$ Hz), 4.47 (d, 1H, H-1, J_{12} =9.4 Hz), 4.16 (dd, 1H, H-3, J_{23} =8.9 Hz, J_{23} = 9.9 Hz), 4.00 (dd, 1H, H-6, $J_{6a,6b} = -11.4$ Hz, $J_{5,6a} =$ 1.9 Hz), 3.83 (m, 2H, H-4, H-5), 3.36 (m, 2H, H-2, H-6b), 3.20, 3.18 (2×s, 6H, OCH₃ BDA), 2.74 (m, 2H, CH₂ SEt), 1.36, 1.28 (2×s, 6H, 2×CH₃ BDA), 1.33 (t, 3H, CH₃ SEt,

 $J=-7.5$ Hz), 1.04 (s, 9H, CH₃ t-Bu Si); ¹³C{¹H} NMR $(CDCl_3)$: δ 138.1 (Cq Bn), 135.5, 135.3 (Cq TBDPS), 135.6, 129.5–126.8 (CH arom), 101.1, 101.0 (2×Cq BDA), 84.1 (C-1), 79.8, 79.0, 77.1, 69.0 (C-2, C-3, C-4, C-5), 74.5 (CH₂ Bn), 63.3 (C-6), 47.8 (2 \times OCH₃ BDA), 26.6 (t-Bu Si), 24.5 (CH₂ SEt), 19.2 (Cq t-Bu Si), 17.8, 17.6 (2×CH₃ BDA), 15.1 (CH₃ SEt). ES-MS: 689 $[M+Na]^+$. HRMS (ES): Calcd C₃₇H₅₁O₇SSi $[M+H]^+$ 667.3135, found 667.3139 (\pm 0.0023).

1,2-O-Isopropylidene-5-O-trityl- α -D-ribofuranoside (20). 1,2-O-Isopropylidene- α -D-ribofuranoside²⁶ (4.8 g, 25.0) mmol) was tritylated as described for the synthesis of 18. Purification of the product was accomplished by column chromatography (Et₂O/light petroleum/Et₃N, 14/85/1, v/v/ v) to give 18 as a white solid (9.13 g, 21.3 mmol, 86%). R_f 0.94 (Et₂O). Mp 101°C. ¹H NMR (CDCl₃): δ 7.58–7.18 (m, 15H, H arom), 5.89 (d, 1H, H-1, J_1 ₂=4.4 Hz), 4.59 (t, 1H, H-2, J_2 ₃=3.7 Hz), 4.17-4.07 (m, 1H, H-4), 4.04-3.87 (m, 1H, H-3), 3.63±3.24 (m, 2H, H-5a, H-5b), 2.30 (d, 1H, OH), 1.57, 1.38 (2×s, 6H, 2×CH₃ isoprop). ¹³C{¹H} NMR (CDCl₃): δ 143.5 (Cq Ph), 128.4–126.7 (CH arom), 112.1 (Cq isoprop), 103.8 (C-1), 86.3 (Cq Tr), 79.4, 78.3, 71.8 (C-2, C-3, C-4), 62.7 (C-5), 26.2 (2 \times CH₃ isoprop). C₂₇H₂₈O₅ (432): Calcd C 74.98, H 6.52; found C 75.14, H 6.59.

General glycosylation procedure

A mixture of acceptor 16, 20 or 21 (1.00 mmol) and donor 18 or $19(1.00-1.25 \text{ mmol})$ was dried by coevaporation with 1,4-dioxane $(3\times5$ mL) and was stirred for 15 min in the appropriate solvent mixture (10 mL) with powdered molecular sieves (4 Å) under an atmosphere of argon. NIS (0.27 g) , 1.20 mmol) and a catalytic amount of TfOH $(9 \mu L,$ 0.11 mmol) were subsequently added. After completion of the reaction (\pm 30 min) as judged by TLC analysis, the reaction mixture was filtered and diluted with EtOAc (15 mL). The filtrate was washed with aq. $Na₂S₂O₃$ (1 M, 5 mL) and sat. aq. NaHCO₃ (5 mL), dried over $MgSO₄$ and concentrated in vacuo to afford the crude product.

 $(2''S,3''S)$ 3-O-(2'-O-Benzyl-3',4'-di-O-(2",3"-dimethoxybutane-2",3"-diyl)-6-O-trityl- α -D-glucopyranosyl)-5,6-di $deoxy-1,2-O-isopropvlidene- α - $D-allo$ -hex-5-enofuranoside$ (22). Glycosylation of 5,6-dideoxy-1,2-O-isopropylidene- α -D-allo-hex-5-enofuranoside (16, 0.93 g, 5.0 mmol) with thioglucoside 17 (4.01 g, 6.0 mmol) was carried out in a mixture of Et_2O/CH_2Cl_2 (50 mL, 10/1, v/v) as described in the general procedure. The product was purified by column chromatography (Et₂O/light petroleum, $1/4 \rightarrow 1/1$, v/v) to give dimer 22 as a white foam. Yield 3.28 g (4.3 mmol, 86%). R_f 0.31 (Et₂O/light petroleum, 1/1, v/v). ¹H NMR (CDCl₃): δ 7.47-7.18 (m, 20H, H arom), 6.01-5.87 (m, 1H, H-5), 5.83 (d, 1H, H-1, $J_{1,2} = 3.7$ Hz), 5.47 (d, 1H, H-6a, $J_{5,6a}$ =16.8 Hz), 5.29 (d, 1H, H-1', $J_{1/2}$ =3.7 Hz), 5.23 (d, 1H, H-6b, $J_{5.6b}$ =10.2 Hz), 4.82 (s, 2H, CH₂ Bn), 4.76 (t, 1H, H-2, $J_{2,3}$ =3.9 Hz), 4.62 (m, 1H, H-3), 4.12 (t, 1H, H-3', $J_{2'3} = J_{3'4'} = 9.4$ Hz), 3.93–3.67 (m, 4H, H-4, H-2', H-4', H-5'), 3.32 (m, 1H, H-6a'), 3.29 (s, 3H, OCH₃ BDA), 3.04 (m, 1H, H-6b'), 3.01 (s, 3H, OCH₃ BDA), 1.57, 1.39, 1.32, 1.16 (4×s, 12H, 4×CH₃ isoprop, BDA); ¹³C{¹H} NMR (CDCl₃): δ 143.5 (Cq Tr), 138.7 (Cq Bn), 134.3 (C-5), 128.3±126.6 (CH arom), 117.5 (C-6), 112.4 (Cq isoprop),

103.7 (C-1), 99.0 (2 \times Cq BDA), 95.2 (C-1'), 86.0 (Cq Tr), 78.4, 77.9, 76.4, 76.0, 69.3, 69.1, 66.2 (C-2, C-3, C-4, C-2', C -3', C -4', C -5'), 71.5 (CH_2 Bn), 61.5 (C -6'), 47.6 ($2 \times OCH_3$ BDA), 27.3 (2×CH₃ isoprop), 17.5, 17.2 (2×CH₃ BDA). ES-MS: 818 [M+Na]⁺. C₄₇H₅₄O₁₁ (795): Calcd C 71.01, H 6.85; found C 70.93, H 6.79.

 $(2''S,3''S)$ 3-O-(2'-O-Benzyl-3',4'-di-O-(2",3"-dimethoxybutane-2",3"-diyl)- α -D-glucopyranosyl)-5,6-dideoxy-1,2- O -isopropylidene- α -D-allo-hex-5-enofuranoside (23). Dimer 22 (0.30 g, 0.38 mmol) was dissolved in a solution of 1% p-TsOH in MeOH/CH₂Cl₂ (7.5 mL, 1/1, v/v) and stirred for 4 h. The mixture was neutralized with Et_3N and concentrated. Purification of the crude product by column chromatography (light petroleum/EtOAc, 1/1, v/v) gave alcohol 23 (0.18 g, 0.32 mmol, 84%) as a white foam. R_f 0.15 (Et₂O/light petroleum, 2/3, v/v); ¹H NMR (CDCl₃): δ 7.43–7.26 (m, 5H, H arom), 5.86–5.78 (m, 1H, H-5), 5.81 (d, 1H, H-1, J_{12} =3.7 Hz), 5.51–5.25 (m, 2H, H-6a, H-6b), 5.13 (d, 1H, H-1', $J_{1',2} = 3.7$ Hz), 4.78 (AB, 2H, CH₂ Bn, $J=-12.4$ Hz), 4.69 (t, 1H, H-2, $J_{2,3}=4.0$ Hz), 4.56 (m, 1H, H-3), 4.15 (t, 1H, H-3', $J_{2/3} = J_{3/4} = 9.6$ Hz), 3.77–3.57 (m, 6H, H-4, H-2', H-4', H-5', H-6a', H-6b'), 3.32, 3.27 (2 \times s, 6H, 2 \times OCH₃ BDA), 2.05 (s, 1H, OH), 1.55, 1.35, 1.32 (3 \times s, 12H, $4 \times CH_3$ isoprop, BDA); ¹³C{¹H} NMR (CDCl₃): δ 138.9 (Cq Ph), 134.3 (C-5), 128.5-127.1 (CH arom), 118.7 (C-6), 112.9 (Cq isoprop), 103.8 (C-1), 99.6, 99.4 $(2 \times Cq \text{ BDA})$, 95.9 $(C-1')$, 78.9, 78.2, 76.5, 76.2, 69.7, 69.0, 66.1 (C-2, C-3, C-4, C-2', C-3', C-4', C-5'), 71.9 $(CH_2$ Bn), 60.9 (C-6'), 48.1, 47.8 (2×OCH₃ BDA), 26.6 $(2 \times CH_3 \text{ isoprop})$, 17.8, 17.6 $(2 \times CH_3 \text{ BDA})$.

 $(2''S,3''S)$ 3-O-(2'-O-Benzyl-3',4'-di-O-(2",3"-dimethoxybutane-2″,3″-diyl)-6′-*O*-trityl-α-D-glucopyranosyl)-1,2-O-isopropylidene-5-O-trityl- α -D-ribofuranoside (24 R=Tr). Condensation of trityl derivative 20 (0.76 g, 1.1 mmol) with 18 (0.47 g, 1.1 mmol) was performed in a mixture of Et_2O CH_2Cl_2 (10 mL, 10/1, v/v) as described in the general glycosylation procedure. The disaccharide was purified by column chromatography (Et₂O/light petroleum, $1/4$, v/v) to give the title compound as a white foam. Yield 0.85 g (0.82 mmol, 75%). R_f 0.68 (Et₂O/light petroleum, 1/1, v/v); ¹H NMR (CDCl₃): δ 7.41–7.05 (H arom), 6.04 (d, 1H, H-1, $J_{1,2}$ =3.7 Hz), 5.31 (d, 1H, H-1', $J_{1',2}$ =3.7 Hz), 4.85 (t, 1H, H-2, $J_{2,3}$ =3.8 Hz), 4.76 (s, 2H, CH₂ Bn), 4.36 (dd, 1H, H-3, $J_{3,4} = 9.5$ Hz), 4.27 (d, 1H, H-4), 3.87 -3.70 (m, 4H, H-2', H-3', H-4', H-5'), 3.57, 3.39 (m, 2H, H-5), 3.27 (m, 1H, H-6a'), 3.15, 3.18 (2 \times s, 6H, 2 \times OCH₃ BDA), 2.84 (m, 1H, H-6b¹), 1.56, 1.43, 1.30, 1.11 (4×s, 4×CH₃ isoprop, BDA); ¹³C{¹H} NMR (CDCl₃): δ 143.5 (Cq Tr), 138.7 (Cq Bn), 112.5 (Cq isoprop), 104.2 (C-1), 99.0 (2 \times Cq BDA), 94.6 (C-1'), 85.8, 85.7 (2×Cq Tr), 77.4, 76.2, 69.5, 68.4, 59.8 (C-2, C-3, C-4, C-2', C-3', C-4', C-5'), 71.7 (CH₂ Bn), 65.5, 59.8 (C-5, C-6'), 47.9, 47.6 (2 \times OCH₃ BDA), 26.5 (2 \times CH₃ isoprop), 17.8, 17.3 (2 \times CH₃ BDA). ES-MS: 1042 $[M+H]^+$, 1064 $[M+Na]^+$. HRMS (ES): Calcd C₆₅H₆₉O₁₂ [M+H]⁺ 1041.4799, found 1041.4794 (\pm 0.0029).

 $(2''S.3''S)$ $3-O-(2/-O-Benzv1-6-O-t-butvldiphenvlsilv1 3^{\prime},4^{\prime}$ -di- O -(2 $^{\prime\prime},3^{\prime\prime}$ -dimethoxybutane-2 $^{\prime\prime},3^{\prime\prime}$ -diyl)- α -D-glucopyranosyl)-1,2-O-isopropylidene-5-O-t-butyldiphenylsilyl- α -D-ribofuranoside (24). Condensation of compound 21 $(0.43 \text{ g}, 1.00 \text{ mmol})$ with thioglucoside 19 $(0.76 \text{ g},$ 1.15 mmol) was accomplished in a mixture of toluene and 1,4-dioxane (10 mL, 3/1, v/v) according to the general glycosylation procedure. Purification was performed by column chromatography (light petroleum/EtOAc, $6/1 \rightarrow 1/1$ 1, v/v) to afford bis-silylated dimer 24 in 74% yield $(0.76 \text{ g}, 0.74 \text{ mmol})$, as a white foam. R_f 0.57 (light petroleum/EtOAc, 5/1, v/v). ¹H NMR (300 MHz, H-H-COSY, CDCl₃): δ 7.76-7.53 (m, 8H, H-arom Ph-Si), 7.43-7.18 (m, 17H, arom Bn, Ph-Si), 5.76 (d, 1H, H-1, $J_{1,2}$ =3.9 Hz), 5.24 (d, 1H, H-1', $J_{1'2'}=3.9$ Hz), 4.80 (s, 2H, CH₂ Bn), 4.64 (t, 1H, H-2, $J_{2,3}$ =4.0 Hz), 4.24 (ddd, 1H, H-4, $J_{3,4}$ =4.2 Hz, $J_{4,5a} = 2.5$ Hz, $J_{4,5b} = 9.2$ Hz), 4.15 (t, 1H, H-3['], $J_{3',4'} =$ $J_{2,3} = 9.9$ Hz), 4.13 (dd, 1H, H-3), 3.92 (dd, H-5a, $J_{5a,5b} =$ -9.8 Hz), 3.88 (t, 1H, H-4', $J_{4'5'}=9.8$ Hz), 3.82–3.66 (m, 4H, H-5b, H-5', H-6'), 3.68 (dd, 1H, H-2'), 3.29, 3.26 (2 \times s, 6H, 2×OCH₃ BDA), 1.52, 1.39, 1.38, 1.33 (4×s, 4×CH₃) isoprop, BDA), 0.99, 0.91 $(2 \times s, 18)$ H, CH₃ t-Bu Si); ${}^{13}C[{^{1}H}]$ NMR (CDCl₃): δ 139.1 (Cq Bn), 135.9, 135.7, 135.5, 135.4 (CH arom Ph-Si), 133.8, 133.4, 133.0 (Cq Ph-Si), 129.6–127.2 (CH arom Bn, Ph-Si), 112.9 (Cq isoprop), 104.3 (C-1), 99.5 (2 \times Cq BDA), 95.3 (C-1'), 79.5, 77.2, 76.6, 72.7, 70.4, 69.5, 65.5 (C-2, C-3, C-4, C- 2^{\prime} , C-3^{\prime}, C-4^{\prime}, C-5^{\prime}), 72.0 (CH₂ Bn), 62.0, 61.5 (C-5, C-6^{\prime}), 48.1, 48.0 (2 \times OCH₃ BDA), 26.9, 26.8 (2 \times CH₃ isoprop, CH₃ t -Bu Si), 19.3, 19.2 (2 \times Cq t -Bu Si), 18.1, 17.8 (2 \times CH₃) BDA). ES-MS: 1056 $[M+Na]^+$. $[\alpha]_D^{20} = +91.0^\circ$ (c 1.0 CHCl₃). C₅₉H₇₆O₁₂Si₂ (1033): Calcd C 68.57, H 7.41; found C 68.75, H 7.49.

 $(2''S,3''S)$ 3-O-(2'-O-Benzyl-6,7-dideoxy-3',4'-di-O-(2",3"dimethoxybutane-2",3"-diyl)- α -D-gluco-hept-6-enopyranosyl)-5,6-dideoxy-1,2-O-isopropylidene- α -D-allo-hex-5-eno**furanoside (26).** To a cooled $(-60^{\circ}C)$ solution of oxalyl chloride (0.38 mL, 4.3 mmol) in CH_2Cl_2 (6.7 mL) under a N_2 atmosphere was added dropwise a solution of DMSO (0.61 mL, 8.3 mmol) in CH_2Cl_2 (3.4 mL). After stirring for 2 min, compound 23 (0.88 g, 1.60 mmol) in CH_2Cl_2 (5 mL) was added dropwise and the reaction mixture was stirred at -60° C for 30 min. Et₃N (2.6 mL, 18.4 mmol) was added and after 10 min the solution was allowed to warm to RT. The mixture was diluted with CH_2Cl_2 (30 mL) and washed with sat. aq. NaCl (2×10 mL). The organic layer was dried over MgSO₄, filtered and concentrated under reduced pressure. The crude aldehyde was immediately used without further purification. To a solution of methyltriphenylphosphonium bromide (0.86 g, 2.4 mmol) in THF (18 mL) under a nitrogen atmosphere was added n -butyllithium (1.5 mL, 1.6 M in hexanes). After stirring for 1 h, a solution of the aldehyde (1.6 mmol) in CH_2Cl_2 (3.5 mL) was added dropwise at 0° C to the yellow suspension. When TLC analysis showed complete conversion into a more lipophilic product the reaction mixture was filtered over silica gel, concentrated and purified by column chromatography $(Et₂O/light$ petroleum, $1/4$, v/v) to afford disaccharide 26 as a colorless oil. Yield 0.62 g (1.12 mmol, 70%). R_f 0.63 (light petroleum/EtOAc, 3/1, v/v); ¹H NMR (CDCl₃): δ 7.44-7.22 (m, 5H, H arom), 5.96-5.76 (m, 2H, H-5, H-6[']), 5.80 (d, 1H, H-1, $J_{1,2}$ =3.7 Hz), 5.52–5.22 (m, 4H, H-6, H-7'), 5.14 (d, 1H, H-1', $J_{1/2}$ = 3.7 Hz), 4.79 (AB, 2H, CH₂ Bn, $J=-12.4$ Hz), 4.71 (dd, 1H, H-2, $J_2=4.4$ Hz), 4.57 (dd, 1H, H-3, $J_{3,4}$ =9.5 Hz), 4.14 (m, 2H, H-4', H-3'), 3.72 (dd, 1H, H-5', $J_{5',6'}=4.4$ Hz, $J_{4',5'}=9.5$ Hz), 3.62 (dd, 1H, H-2', $J_{2',3'}=9.3$ Hz), 3.42 (t, 1H, H-4', $J_{3',4'}=9.5$ Hz),

3.32, 3.21 (2£s, 6H, 2£OCH3 BDA), 1.56, 1.36, 1.35, 1.31 $(4 \times s, 12H, 4 \times CH_3 \text{ isoprop}, BDA);$ ¹³C{¹H} NMR (CDCl₃): δ 138.9 (Cq Ph), 134.3 (C-5), 133.6 (C-6'), 128.0–127.2 (CH arom), 118.6 (C-7'), 118.0 (C-6), 112.9 (Cq isoprop), 103.8 (C-1), 99.6, 99.3 (2 \times Cq BDA), 95.7 (C-1'), 78.8, 78.2, 76.6, 76.2, 70.4, 69.8, 69.2 (C-2, C-3, C-4, C-2', C-3', C-4', C-5'), 71.5 (CH₂ Bn), 48.1, 47.7 (2 \times OCH₃ BDA), 26.6 (2×CH₃ isoprop), 17.8, 17.6 (2×CH₃ BDA); ES-MS: 566 $[M+NH_4]^2$, 571 $[M+Na]^2$, 587 $[M+K]^2$. HRMS (ES): Calcd $C_{29}H_{41}O_{10}$ $[M+H]^{\dagger}$ 549.2710, found 549.2714 (\pm 0.0019).

 $(2''S,3''S)$ 3-O-(6'-O-Allyl-2'-O-benzyl-3',4'-di-O-(2",3"dimethoxybutane- $2^{\prime\prime},3^{\prime\prime}$ -diyl)- α -D-glucopyranosyl)-5,6dideoxy-1,2-O-isopropylidene- α -D-allo-hex-5-enofuranoside (27). To a cooled solution (0°C) of dimer 23 (0.20 g, 0.36 mmol) in DMF (3 mL) was added sodium hydride (60% wt, 22 mg, 0.54 mmol). After stirring for 15 min, allyl bromide $(34 \mu L, 0.40 \text{ mmol})$ was added and the reaction mixture allowed to stir for 3 h after which it was quenched with MeOH (0.3 mL). The mixture was diluted with Et_2O (5 mL) H_2O (15 mL) was added and the DMF/ $H₂O$ layer was extracted with Et₂O. The organic layer was rinsed with H₂O ($2\times$ 5 mL), dried over MgSO₄, filtered and concentrated in vacuo. After purification by column chromatography (Et₂O/light petroleum, $1/3$, v/v) compound 27 was obtained as a white foam. Yield 0.15 g (0.25 mmol, 70%). R_f 0.41 (Et₂O/light petroleum, 1/1, v/v); ¹H NMR (CDCl₃): δ 7.43–7.24 (m, 5H, H arom), 5.95–5.73 (m, 2H, H-5, CH allyl), 5.79 (d, 1H, H-1, $J_{1,2}$ =3.7 Hz), 5.51– 5.18 (m, 4H, H-6, CH_2 =CH allyl), 5.14 (d, 1H, H-1', $J_{1/2}$ = 4.4 Hz), 4.78 (AB, 2H, CH₂ Bn, J = -12.4 Hz), 4.70 $(t, 1H, H-2, J_{2,3}=3.9 \text{ Hz})$, 4.55 (m, 1H, H-3), 4.17-3.97 (m, $3H, H-4, H-3', OCH₂$ allyl), $3.80-3.60$ (m, 6H, H-2', H-4', $H-5'$, $H-6'$, OCH₂ allyl), 3.31, 3.25 (2×s, 6H, 2×OCH₃) BDA), 1.55, 1.35, 1.33, 1.31 (4×s, 12H, 4×CH₃ isoprop, BDA); ${}^{13}C({}^{1}H)$ NMR (CDCl₃): δ 138.9 (Cq Bn), 134.4, 134.3 (C-5, OCH₂ allyl), 128.0, 127.2 (CH arom Bn), 118.5 (C-6), 116.9 (CH₂=CH allyl), 112.8 (Cq isoprop), 103.8 (C-1), 99.5, 99.3 (2 \times Cq BDA), 95.8 (C-1[']), 78.7, 78.2, 76.5, 76.0, 69.2, 69.0, 65.8 (C-2, C-3, C-4, C-2', C-3', C-4', C-5'), 72.4, 71.9 (CH₂ Bn, CH allyl), 67.5 $(C-6')$, 48.0, 47.8 (2 \times OCH₃ BDA), 26.6 (2 \times CH₃ isoprop), 17.8, 17.6 (2 \times CH₃ BDA); ES-MS: 594 $[M+H]^+$, 516 $[M+Na]^+$. HRMS (ES): Calcd C₃₁H₄₅O₁₁ $[M+H]^+$ 593.2972, found 593.2969 (\pm 0.0017).

 $(2''S,3''S)$ 5-O-Allyl-3-O-(6'-O-allyl-2'-O-benzyl-3',4'-di- $O-(2^n,3^n$ -dimethoxybutane- $2^n,3^n$ -diyl)- α -D-glucopyranosyl)-1,2-O-isopropylidene- α -D-ribofuranoside (28). Compound 24 (1.14 g, 1.10 mmol) was dissolved in a mixture of 1,4-dioxane (15 mL) and a 1.0 M solution of TBAF in THF (3.29 mL) . After stirring overnight at 50° C the mixture was concentrated and the oily residue was dissolved in EtOAc (25 mL). The solution was rinsed with sat. aq. NaCl $(3\times10 \text{ mL})$ and H₂O (10 mL). The organic layer was dried $(MgSO₄)$ and concentrated. Purification was effected by column chromatography (EtOAc/MeOH, $1/0 \rightarrow 98/2$, v/v) to give diol 25 as a white foam. Yield 0.59 g (1.04 mmol, 95%). R_f 0.18 (EtOAc). Allylation of compound 25 (0.57 g, 1.02 mmol) was executed as described for the preparation of 27. The crude product was purified by column chromatography (Et₂O/light petroleum, $1/6$, v/v), to afford 28 as a

colorless oil. Yield 0.59 g (0.93 mmol, 89%). R_f 0.90 (EtOAc); ¹H NMR (300 MHz, H-H-COSY, CDCl₃): δ 7.42–7.24 (H arom), 5.97–5.83 (m, 2H, 2 \times CH allyl), 5.80 (d, 1H, H-1, $J_{1,2}$ =3.7 Hz), 5.31–5.13 (m, 4H, 2 \times CH₂CH allyl), 5.18 (d, 1H, H-1', $J_{1/2} = 4.4$ Hz), 4.77 (AB, 2H, CH₂ Bn, J=-12.4 Hz), 4.70 (t, 1H, H-2, $J_{2,3}=4.0$ Hz), 4.29 (ddd, H-4, $J_{3,4}$ =9.3 Hz, $J_{4,5a}$ =1.9 Hz, $J_{4,5b}$ =4.0 Hz), $4.18-3.92$ (m, 8H, H-3, H-3', H-4', H-5', 2 \times OCH₂ allyl), 3.81 (m, 1H, H-6a'), 3.75 (dd, 1H, H-5a, $J_{5a,5b} = -11.2$ Hz), $3.65-3.51$ (m, $3H$, $H-2'$, $H-6b'$, $H-5b$), $3.32, 3.25$ ($2 \times s$, $6H$, 2×OCH₃ BDA), 1.53, 1.35, 1.31 (3×s, 12H, 4×CH₃ isoprop, BDA). ¹³C{¹H} NMR (CDCl₃): δ 138.7 (Cq Bn), 134.4, 134.2 (2×CH allyl), 127.8, 127.1 (CH arom Bn), 116.9 $(2 \times CH_2=CH$ allyl), 112.7 (Cq isoprop), 104.0 (C-1), 99.4, 99.3 (2×Cq BDA), 95.6 (C-1'), 77.4, 76.3, 75.9, 73.1, 69.2, 68.9, 65.5 (C-2, C-3, C-4, C-2', C-3', C-4', $(C-5')$, 72.3, 71.8 (CH₂ Bn, OCH₂ allyl), 67.8, 67.2 (C-6['], C-5), 48.0, 47.7 (2 \times OCH₃ BDA), 26.5 (2 \times CH₃ isoprop), 17.5, 17.5 (2×CH₃ BDA); ES-MS: 659 [M+Na]⁺. [α]²⁰= $+135.4^{\circ}$ (c 2.0 CHCl₃). HRMS (ES): Calcd C₃₃H₄₉O₁₂ $[M+H]^+$ 637.3223, found 637.3219 (\pm 0.0027). C₃₃H₄₈O₁₂ (636): Calcd C 62.25, H 7.60; found C 62.45, H 7.64.

General procedure for ring-closing metathesis

Residual H_2O was removed from the diene (1.0 mmol) by coevaporation with dry toluene $(3\times5 \text{ mL})$, after which it was dissolved in dry toluene (40 mL). The solution was degassed by bubbling through with argon for 20 min. Catalyst 29 (41 mg, 5 mol%) was added and degassing was continued for 20 min, after which the solution was stirred overnight under an argon atmosphere. When TLC analysis indicated termination of the reaction, the reaction mixture was concentrated and the product was purified by column chromatography (light petroleum/Et₂O, $8/1 \rightarrow 1/1$, v/v).

Dimer of 27. Attempted ring-closing metathesis of diene 27 (100 mg, 0.17 mmol) according to the general procedure yielded a single isomer of dimer 30 as a brownish oil. Yield 66 mg (0.12 mmol, 69%). R_f 0.83 (Et₂O); ¹H NMR (CDCl₃): δ 7.43–7.27 (m, 10H, H arom Bn), 5.89–5.73 (m, 4H, H-5, CH allyl), 5.80 (d, 2H, H-1, J_{12} =3.7 Hz), 5.50– 5.23 (m, 4H, H-6a, H-6b), 5.14 (d, 2H, H-1', $J_{1'2} = 3.7$ Hz), 4.77 (AB, 4H, CH₂ Bn, $J=-12.4$ Hz), 4.69 (t, 2H, H-2, $J_{2,3}=3.9$ Hz), 4.57 (dd, 2H, H-3), 4.09 (t, 2H, H-3', $J_{3',4'}=J_{2',3'}=9.7$ Hz), 4.00–3.98 (m, 4H, H-4, OCH₂ allyl), $3.78-3.43$ (m, 12H, H-2', H-4', H-5', H-6a', H-6b', OCH₂ allyl), 3.31, 3.23 (2×s, 12H, 2×OCH₃ BDA), 1.55, 1.35, 1.34, 1.30 (4 \times s, 24H, CH₃ isoprop, BDA); ¹³C{¹H} NMR (CDCl3): ^d 138.9 (Cq Ph), 134.4 (C-5, CH allyl), 128.0, 127.6, 127.2 (CH arom Bn), 118.6 (C-6), 112.9 (Cq isoprop), 103.8 (C-1), 99.6, 99.3 (2×Cq BDA), 95.9 (C-10), 79.6, 78.7, 76.5, 76.0, 69.2, 69.0, 65.8 (C-2, C-3, C-4, C-2', C-3', C-4', C-5'), 72.4 (OCH₂ allyl), 71.3 (CH₂ Bn), 67.6 (C-6'), 48.1, 47.8 (2 \times OCH₃ BDA), 26.6 (2 \times CH₃ isoprop), 17.8, 17.7 (2 \times CH₃ BDA). ES-MS: 1158 $[M+H]^+$, 1170 $[M+Na]^+$. HRMS (ES): Calcd C₆₀H₈₅O₂₂ $[M+H]$ ⁺ 1157.5542, found 1157.5549 (\pm 0.0028).

 $(E/Z, 2''S, 3''S)$ 5,6'-Di-O-but-2-en-1,4-diyl-3-O-(2'-Obenzyl-3',4'-di-O-(2",3"-dimethoxybutane-2",3"-diyl)- α d-glucopyranosyl)-1,2-O-isopropylidene-a-d-ribofuranoside (31). Ring-closing metathesis of diene 28 (0.30 g,

0.47 mmol) was effected according to the general procedure to give a 2:1 mixture of E/Z-isomers of 14-membered macrocycle 31. The light brownish oily product was used in the next reaction without purification. Combined yields 0.21 g (0.32 mmol, 75%). R_f 0.53 and 0.42 (Et₂O). The higher-running product was the major isomer: ^IH NMR (600 MHz, H-H-COSY, CDCl₃): δ 7.44–7.23 (m, 5H, H arom Bn), 5.89 (m, 2H, H-b, H-c), 5.78 (d, 1H, H-1, $J_{1,2} = 3.7$ Hz), 5.12 (d, 1H, H-1', $J_{1',2'} = 4.2$ Hz), 5.78 (AB, 2H, CH₂ Bn, J=-12.1 Hz), 4.69 (t, 1H, H-2', J_{2,3}=4.0 Hz), 4.52 (dd, 1H, H-3, $J_{3,4}$ =9.1 Hz), 4.27 (m, 2H, H-4, H-a), 4.17 (m, 1H, H-d), 4.14 (t, 1H, H-3', $J_{2',3'}=J_{3',4'}=9.8$ Hz), 4.06 (dt, 1H, H-5', $J_{4',5'}=J_{5',6b'}=9.9$ Hz, $J_{5',6a'}$ 1.5 Hz), 3.95 (dd, 1H, H-d', $J_{c,d} = 5.3$ Hz, $J_{d,d} = -10.2$ Hz), 3.89 (dd, 1H, H-a', $J_{a',b}$ =5.3 Hz, $J_{a,a'}$ =-9.9 Hz), 3.83 (d, 1H, H-5a, $J_{5a,5b} = -11.2 \text{ Hz}$, 3.75 (dd, 1H, H-6a', $J_{6a,6b} = -12.0 \text{ Hz}$), 3.67 (dd, 1H, H-5b, $J_{4,5b}$ =1.8 Hz), 3.64 (dd, 1H, H-2⁷), 3.59 (dd, 1H, H-6b'), 3.42 (t, 1H, H-4'), 3.33 , 3.21 ($2 \times s$, 6H, 2×OCH₃ BDA), 1.50, 1.35, 1.34, 1.32 (4×s, 12H, $4 \times CH_3$ isoprop, BDA); ¹³C{¹H} NMR (CDCl₃): δ 139.0 (Cq Ph), 130.5, 129.9 (C-b, C-c), 128.0-127.0 (CH arom), 112.8 (Cq isoprop), 104.2 (C-1), 99.7, 99.4 (2×Cq BDA), 95.4 (C-1'), 77.6, 76.2, 71.9, 69.5, 67.7, 67.4 (C-2, C-3, C-4, C-2', C-3', C-4', C-5'), 71.8, 69.4, 67.2, 66.4, 66.0 (C-a, C-d, C-5, C-6', CH₂ Bn), 48.3, 47.8 (2 \times OCH₃ BDA), 26.8, 26.6 (2×CH₃ isoprop), 17.9, 17.7 (2×CH₃ BDA); ES-MS: 632 $[M+Na]^+$. C₃₁H₄₄O₁₂ (609): Calcd C 61.17, H 7.29; found C 61.11, H 7.35.

 $(2''S,3''S)$ 3-O-(2'-O-Benzyl-3',4'-di-O-(2",3"-dimethoxybutane-2″,3″-diyl)- α -D-glucopyranosyl)-5,6′-di- O -(butane-1,4-diyl)-1,2-O-isopropylidene- α -D-ribofuranoside (32). A mixture of E/Z isomers of 31 (0.21 g, 0.32 mmol) was dissolved in EtOAc (10 mL) and the solution was degassed and placed under a blanket of N_2 . A catalytic amount of PtO₂ (7 mg, 10 mol%) was added, the mixture was degassed once more and stirred under a H_2 -atmosphere. TLC analysis revealed complete conversion of 31 after 25 min into a single higher-running product. The reaction mixture was filtered over Glass Fiber (GF/2A, Whatman[®]) and concentrated in vacuo to give 32 in quantitative yield (0.21 g, 0.32 mmol). R_f 0.64 (Et₂O) as a brownish oil. ¹H NMR (CDCl₃, 300 MHz, H-H-COSY): δ 7.43-7.22 (m, 5H, H arom), 5.83 (d, 1H, H-1, $J_{1,2} = 3.8$ Hz), 5.17 (d, 1H, H-1', $J_{1/2}$ = 4.2 Hz), 4.79 (AB, 2H, CH₂ Bn, J = -12.4 Hz), 4.74 (dd, 1H, H-2, $J_{2,3}$ =4.5 Hz), 4.52 (dd, 1H, H-3, $J_{3,4}$ =9.1 Hz), 4.25 (dd, 1H, H-4, $J_{4,5a}$ =1.4 Hz), 4.15–4.04 (m, 3H, H-3', H-5[']), 3.86 (dd, 1H, H-5a, $J_{5a,5b}$ =-11.6 Hz), 3.74 (dd, 1H, H -6a', $J_{6a',6b'} = -11.7$ Hz, $J_{5',6b'} = 1.6$ Hz), 3.65 (dd, 1H, $H-2'$, $J_{2',3'}=10.1$ Hz), $3.61-3.41$ (m, 7H, H-5b, H-4', H_0 -6b', OCH₂ n-Bu), 3.33, 3.21 (2×s, 6H, 2×OCH₃ BDA), 1.88, 1.63 (m, 4H, CH₂ n-Bu), 1.49, 1.35, 1.32 (4×s, 12H, $4 \times CH_3$ isoprop, BDA); ¹³C{¹H} NMR (CDCl₃): δ 138.8 (Cq Ph), 127.9±126.9 (CH arom), 112.7 (Cq isoprop), 104.0 (C-1), 99.5, 99.3 (2 \times Cq BDA), 95.2 (C-1'), 78.0, 76.2, 71.9, 69.6, 67.3 (C-2, C-3, C-4, C-2', C-3', C-4', $(C-5')$, 71.8, 70.0, 69.1, 68.7, 67.0 (2 \angle OCH₂ n-Bu, C-5, C -6', CH_2 Bn), 48.1, 47.7 (2 \times OCH₃ BDA), 26.6 (2 \times CH₃) isoprop), 25.8, 25.5 (2 \times CH₂ n-Bu), 17.7, 17.5 (2 \times CH₃ BDA). $\left[\alpha\right]_D^{20} = +138.6^\circ$ (c 1.0 CHCl₃). ES-MS: 634 [M+ Na]⁺. HRMS (ES): Calcd C₃₁H₄₇O₁₂ [M+H]⁺ 611.3067, found 611.3072 (\pm 0.0019). Calcd C 57.69, H 6.45; found C 57.82, H 6.53.

1,2-Di-*O*-acetyl-3-*O*-(3',4'-di-*O*-acetyl-2'-*O*-benzyl-β-Dglucopyranosyl)-5,6′-di-*O*-(butane-1,4-diyl)-D-ribofuranoside (33). Compound 32 (0.15 g, 0.24 mmol) was dissolved in a solution of AcOH/H₂O/(CH₂OH)₂, $14/6/3$, $v/v/v$ (7 mL) and was refluxed for 90 min. TLC analysis (MeOH/EtOAc, 8/92, v/v) indicated the appearance of one product $(R_f 0.42)$. The reaction mixture was concentrated under reduced pressure, coevaporated with toluene $(3\times5$ mL) and stirred in a mixture of in pyridine/acetic anhydride (8 mL, 2/1, v/v). The mixture was concentrated, coevaporated with toluene $(3\times5$ mL) and subjected to column chromatography (light petroleum/EtOAc, $1/0$ \rightarrow 1/1, v/v) to yield 33 (138 mg, 0.22 mmol, 92%, $\alpha:\beta \sim 1:9$ as a white foam. R_f 0.15 (EtOAc/light petroleum, $1/1$, v/v). ${}^{13}C[{^1}H]$ NMR $(CDCl_3)$: δ 169.9 (C=O Ac), 138.5 (Cq Bn), 128.2, 127.7, 127.4 (CH arom Bn), 97.7 (C-1), 94.9 (C-1'), 80.7, 76.7, 71.6, 70.9, 70.3, 69.6, 68.7 (C-2, C-3, C-4, C-2', C-3', C-4', C-5'), 77.6, 76.4, 72.3, 69.9, 76.6 $(2 \times OCH_2 n$ -Bu, C-5, C-6', CH₂ Bn), 26.3 (2 \times CH₂ n-Bu), 20.6 (4 \times CH₃ Ac). ES-MS: 565 $[M-OAc]$ ⁺, 647 $[M+Na]$ ⁺, 1272 $[2M-Na]$ ⁺. HRMS (ES): Calcd C₃₀H₄₁O₁₄ [M+H]⁺ 625.2506, found 625.2508 $(\pm 0.0016).$

2'-O-Acetyl-6-N-benzoyl-5',6"-di-O-(n-butane-1,4-diyl)-3'-O-(3",4"-di-O-acetyl-2"-O-benzyl-α-d-glucopyranosyl)adenosine (34). A suspension of 6-N-benzoyladenine $(0.14 \text{ g}, \ 0.60 \text{ mmol})$ in $1,1,1,3,3,3$ -hexamethyldisilazane (1.1 mL) and pyridine (0.5 mL) was refluxed for 7 h under an Ar atmosphere. The reaction mixture was cooled, diluted with toluene (5 mL) and concentrated in vacuo under careful exclusion of H_2O . The residual oil was diluted with toluene $(3\times5$ mL) and concentrated in vacuo to remove excess 1,1,1,3,3,3-hexamethyldisilazane. Tetraacetate 33 $(0.13 \text{ g}, 0.20 \text{ mmol})$ in $(CH_2Cl)_2$ (5 mL) and a catalytic amount of TMSOTf $(8 \mu L, 25 \text{ mol\%)}$ were added to the silylated 6-N-benzoyladenine. After stirring for 16 h at reflux temperature TLC analysis showed conversion of 33 into a lower-running product. The reaction mixture was quenched with Et₃N (0.25 mL), diluted with CH_2Cl_2 (10 mL) and poured into sat. aq. NaHCO₃ (5 mL). The organic phase was washed with H_2O (5 mL), dried $(MgSO₄)$, filtered and concentrated under reduced pressure. The crude product was purified by column chromatography $(CH_2Cl_2/MeOH$, 1/0 to 95/5, v/v). Concentration of the appropiate fractions yielded 34 as a yellowish foam. Yield 0.13 g (0.16 mmol, 82%); R_f 0.65 (EtOAc/MeOH, 95/5, v/v); ¹H NMR (CDCl₃, 600 MHz, H-H-COSY): δ 9.05 (s, 1H, NH), 8.81 (s, 1H, H-2), 8.63 (s, 1H, H-8), 8.02, (d, 2H, arom Bz, $J=7.4$ Hz), 7.60 (t, 1H, arom Bz, $J=7.4$ Hz), 7.52 $(t, 2H, \text{arom Bz}, J=7.8 \text{ Hz}), 7.38-7.28 \text{ (m, 5H, H areb)}$, 6.27 (d, 1H, H-1', $J_{1',2'}=1.8$ Hz), 5.88 (dd, 1H, H-2', $J_{2',3'}=$ 4.7 Hz), 5.44 (t, 1H, H-3ⁿ, $J_{2^n,3^n} = J_{3^n,4^n} = 9.6$ Hz), 5.07 (d, 1H, H-1ⁿ, $J_{1^n,2^n} = 3.7$ Hz), 4.86 (dd, 1H, H-3¹, $J_{3^1,4^1} =$ 7.9 Hz), 4.71 (t, 1H, H-4", $J_{4", 5"} = 9.8$ Hz), 4.60 (AB, 2H, CH₂ Bn, $J=-11.9$ Hz), 4.48 (m, 1H, H-4'), 4.01 (dd, 1H, H -5a', $J_{5a',5b'} = -11.2$ Hz, $J_{4',5a'} = 2.7$ Hz), 3.97 (dt, 1H, $H-5^{\prime\prime}$, $J_{5^{\prime\prime},6^{\prime\prime}}=2.0$ Hz), 3.71 (dd, 1H, H-5b['], $J_{4^{\prime},5b^{\prime}}=2.7$ Hz), 3.69±3.67, 3.64±3.60 (2£m, 2H, H-a/d), 3.55 (1H, dd, H-2", $J_{2}y_1 = 10.0$ Hz), 3.50–3.42 (m, 2H, H-6a", H-a/d), 3.39 (1H, dd, H-6bⁿ, $J_{6a''.6b''} = -11.0$ Hz), 2.04, 1.92, 1.87 $(3x, 9H, CH, 4c), 1.94-1.82, 1.74-1.69$ (2 x m, 4H, H-b, H-c); ${}^{13}C[{^1}H]$ NMR (150 MHz, C-H-COSY, CDCl₃): δ 170.1, 170.0, 168.8 $(3 \times C = 0 \text{ Ac})$, 164.6 $(C=0 \text{ Bz})$,

152.3, 151.1, 149.4, 141.8 (C-4, C-6, C-2, C-8), 137.6 (Cq Bn), 133.7 (Cq Bz), 132.7 (CH arom Bz), 128.3, 128.2, 127.9, 127.6 (CH arom), 123.5 (C-5), 95.9 (C-1ⁿ), 87.8 (C-1'), 81.3 (C-4'), 76.9 (C-2"), 73.8 (CH₂ Bn), 72.8 (C- $2'$), 71.7 (C-3"), 71.4 (C-3'), 71.1, 70.7 (OCH₂ n-Bu), 70.0 $(C-6'')$, 69.9 $(C-5'')$, 69.8 $(C-4'')$, 68.0 $(C-5')$, 26.7, 26.3 (CH₂ n-Bu), 20.8, 20.7, 20.4 (3×CH₃ Ac); ES-MS: 827 $[M+Na]^+,$ 805 $[M+H]^+,$ 565 $[M-A^{Bz}]+$, 240 $[A^{Bz}+]$ H ⁺. [α] $_{\text{D}}^{20}$ = +43.8° (c 1.0 CHCl₃). HRMS (ES): Calcd $C_{40}H_{46}N_5O_{13}$ [M+H]⁺ 804.3091, found 804.3082 (\pm 0.0029). Calcd C 59.77, H 5.64; found C 59.85, H 7.58.

 $5', 6''$ -Di- O -(n-butane-1,4-diyl)-3'- O -(α -D-glucopyranosyl)adenosine 2′,3″,4″-trisphosphate—Cyclophostin (14). To a solution of glucopyranosyl adenosine 34 (83 mg, 0.10 mmol) in 1,4-dioxane (3 mL) was added a solution of KO-tBu in MeOH (1 M, 4 mL). After stirring vigorously for 1 min, the reaction mixture was neutralized by the addition of AcOH (0.23 mL, 4.0 mmol). The solution was poured into sat. aq. NaHCO₃ (5 mL) and the resulting mixture which was extracted with CH_2Cl_2 (2×10 mL). The organic phase was washed with $H_2O(3 \text{ mL})$, dried (MgSO₄), filtered and concentrated in vacuo. The crude deacetylated dimer was used without purification. R_f 0.43 (EtOAc/MeOH, 9/1, v/v). ES-MS: 679 $[M+H]^{+}$, 701 $[M+Na]^{+}$. A mixture of the triol $(70 \text{ mg}, 0.10 \text{ mm})$ and N,N-diisopropylaminobis-[2-(methylsulfonyl)ethyl]-phosphine (0.23 g, 0.62 mmol) was coevaporated with 1,4-dioxane $(2\times5$ mL). The mixture was dissolved in CH_2Cl_2 (3 mL) and a solution of 1H-tetrazole (56 mg, 0.80 mmol) in CH_3CN (3 mL) was added. TLC analysis indicated the formation of a single product $(R_f \ 0.43, \ CH_2Cl_2/MeOH, \ 9/1, \ v/v)$ which was complete after 30 min. Oxidation of the intermediate phosphite triesters was effected by the addition of t-BuOOH (0.80 mL) at 0° C. After 30 min the mixture was diluted with CH_2Cl_2 (10 mL), washed with H₂O (5 mL) and dried (MgSO4). Concentration of the organic phase gave trisphosphate 35. R_f 0.14 (CH₂Cl₂/MeOH, 9/1, v/v). ${}^{31}P\{{}^{1}H\}$ NMR (CDCl₃): δ -2.00, -2.39, -2.45. Crude 35 was dissolved in a mixture of NaOH (4 M)/1,4-dioxane/MeOH (1/14/5, v/v/v, 10 mL) and stirred for 16 h. The mixture was neutralized with AcOH (0.06 mL), concentrated and the product was purified by gel-filtration over a Fractogel HW-40 column (elution: 0.15 M triethyl ammonium carbonate). Concentration, coevaporation with $MeOH/H₂O$ $(3\times5 \text{ mL}, 4/1, v/v)$ and lyophilization of the appropriate fractions afforded the $2/-O$ -benzyl derivative of cyclophostin in pure form. Yield 82 mg (56% over the three steps from 34). Debenzylation was accomplished by dissolving the compound in $H_2O(10 \text{ mL})$ and stirring the solution in the presence of Pd-black (50 mg) and AcOH (2 drops) under a H_2 atmosphere. After 16 h the catalyst was removed by filtration over Glass Fiber (GF/2A, Whatman[®]). The filtrate was concentrated and lyophilized to give cyclophostin 14 which was converted into the $Na⁺$ -form by ionexchange with $Dowex^{\circ}$ 50Wx4 (Na⁺-form) and imino diacetate resin (Chelex®, Na⁺ form). Lyophilization gave 14 as a white fluffy foam. Yield 45 mg (48 μ mol). ¹H NMR $(D_2O, 600 \text{ MHz}, H-H-COSY)$: δ 8.39 (H-2), 8.25 (H-8), 6.44 (d, 1H, H-1', $J_{1/2}$ = 2.3 Hz), 5.32 (d, 1H, H-1'', $J_{1'',2''}$ =4.1 Hz), 5.29 (ddd, 1H, H-2', $J_{2',3}$ =5.0 Hz, ${}^{3}J_{p}$ = 7.9 Hz), 4.95 (dd, 1H, H-3', $J_{3',4'}=5.9$ Hz), 4.47 (m, 1H, H-4'), 4.28 (q, 1H, H-3", $J_{2^{\prime\prime},3^{\prime\prime}} = J_{3^{\prime\prime},4^{\prime\prime}} = {}^3J_P = 9.2$ Hz), 4.11

(d, 1H, H-6a", J=-11.7 Hz), 4.00 (t, 1H, H-5", $J_{4''.5''}$ = $J_{5^{\prime\prime}6h^{\prime\prime}}$ =9.8 Hz), 3.86–3.81 (m, 2H, H-4", H-2"), 3.77 (dd, 1H, H-5a', $J_{4',5a'}=3.9$ Hz, $J_{5a',5b'}=-11.7$ Hz), $3.69-3.61$ (m, $3H$, OCH₂ n-Bu, H-6bⁿ, H-5bⁿ), $3.57-3.51$, $3.47-3.44$ $(2\times m, 3H, OCH_2 n-Bu), 1.68-1.63, 1.56-1.52, 1.50-1.42$ $(3 \times m, 4H, 2 \times CH_2 n-Bu);$ ¹³C{¹H} NMR (D₂O): δ 148.1 (C-4/C-6), 153.6 (C-2), 120.1 (C-5), 96.4 (C-1"), 88.2 (C-1'), 81.9 (C-4'), 77.3, 74.0, 73.3, 72.8, 71.1 (C-2', C-3', C-2", C-3", C-4", C-5"), 70.7, 70.3, 70.0, 68.4, 69.5 (2×OCH₂ n-Bu, C-5', C-6"), 26.3, 25.9 (2 \times CH₂ n-Bu); ³¹P NMR (D₂O, 242 MHz, P–H-COSY): δ 4.44 (P-3", P-2'), 2.92 (P-4"); ES-MS: 722 $[M-H]$, 360.5 $[M-2H]^{2-}$, 691 $[M-3H+2Na]$, 713 $[M-4H+3Na]$. HRMS (ES): Calcd $C_{20}H_{31}N_5O_{18}P_3$ [M-H]^{-722.0877}, found 722.0865 (± 0.0025) .

Propargyl 2'-O-acetyl-3'-O-(3",4"-di-O-acetyl-2"-O-benzyl- α -D-glucopyranosyl)-5′,6″-di- O -(butane-1,4-diyl)-β-**D-ribofuranoside (36).** Compound 33 (0.13 g, 0.21 mmol) was coevaporated with $(CH_2Cl)_2$ and dissolved in $(CH_2Cl)_2$ (3 mL) . Propargyl alcohol $(18 \mu L, 0.31 \text{ mmol})$ and activated molecular sieves (4 Å) were added and the mixture was stirred for 15 min under a blanket of N_2 . TMSOTf $(8 \mu L, 25 \text{ mol})\%$ was added and after 6 h the reaction mixture was filtered over $Hyflo^{\circledast}$, diluted with Et₂O (10 mL) and washed with sat. aq. NaHCO₃. The organic layer was dried over $MgSO₄$ and concentrated in vacuo. The crude residue was purified by column chromatography (light petroleum/EtOAc, $3/1 \rightarrow 0/1$, v/v) to afford acetylene derivative 34 as a colorless oil (70 mg, 0.11 mmol, 54%). R_f 0.38 (EtOAc/light petroleum, 1/1, v/v). ¹H NMR (CDCl₃): δ 7.33–7.29 (m, 5H, H-arom Bn), 5.39 (t, 1H, H-3", $J_{2^{\prime\prime},3^{\prime\prime}}=J_{3^{\prime\prime},4^{\prime\prime}}=9.5$ Hz), 5.36 (d, 1H, H-2', $J_{2^{\prime},3}$ = 4.4 Hz), 5.16 (s, 1H, H-1'), 5.02 (d, 1H, H-1", $J_{1^{\prime\prime},2^{\prime\prime}}$ =3.6 Hz), 4.76–4.69 (m, 2H, H-4^{$\prime\prime$}, H-5^{$\prime\prime$}), 4.59 (AB, 2H, CH₂ Bn, J=-11.4 Hz), 4.32-4.22 (m, 3H, H-3['], H-1 propargyl), 3.97 (ddd, 1H, H-4', $J_{4',5'}=8.6$ Hz, $J_{4',5b'}=$ 1.1 Hz), 3.77 (dd, 1H, H-5a', $J_{5a',5b'} = -11.7$ Hz, $J_{4,5a}$ = 2.9 Hz), 3.69–3.30 (m, 7H, H-5b⁷, 2×OCH₂ n-Bu, H-6"), 2.03, 1.92, 1.88 (3 \times s, 9H, CH₃ Ac). 1.95-1.51 (m, 4H, $2 \times CH_2$ n-Bu); ¹³C{¹H} NMR (CDCl₃): δ 170.0 $(3 \times C(O)$ Ac), 137.5 (Cq Ph), 128.2-127.5 (CH arom), 102.2 (C-1'), 95.5 (C-1"), 80.3, 76.7, 73.2, 71.6, 69.8, 69.2 (C-2', C-3', C-4', C-2", C-3", C-4", C-5", CH propargyl), 73.1, 70.8, 70.6, 70.1, 68.6 $(2 \times OCH_2 n-Bu, C-5)$ $C-6$ ⁿ, CH₂ Bn, Cq propargyl), 53.3 (CH₂ propargyl), 26.5, 25.9 (2 \times CH₂ n-Bu), 20.6 (3 \times CH₃ Ac). [α] $_{\text{D}}^{20}$ = +61.3° (c 1.0) CHCl₃). HRMS (ES): Calcd C₃₁H₄₁O₁₃ [M+H]⁺ 621.2547, found 621.2543 (\pm 0.0021). Calcd C 59.99, H 6.50; found C 60.08, H 6.57.

n-Propyl 5′,6″-di-*O-(n-*butane-1,4-diyl)-3′-*O-(*α-ɒ-gluco $pyranosyl$)- β - D -ribofuranoside $2^{\prime},3^{\prime\prime}$,4"-trisphosphate (15). Triacetate 36 (70 mg, 0.11 mmol) was dissolved in a 0.1 M solution of NaOMe in MeOH. After 1 h the mixture was neutralized with Dowex[®] H⁺ which was subsequently removed by filtration. The filtrate was concentrated and the residual oil was dried by coevaparation with 1,4-dioxane $(3\times5 \text{ mL})$. The thus obtained triol $(56 \text{ mg}, 0.11 \text{ mmol})$ and dibenzyl N,N-diisopropyl phosphoramidite (0.14 mL, 0.68 mmol) were dissolved in (CH_2Cl) , (4 mL) and a solution of $1H$ -tetrazole (65 mg, 0.96 mmol) in CH₃CN (2 mL) was added under N_2 . After strirring for 30 min TLC analysis $(Et₂O/light$ petroleum, $1/1$, v/v) showed complete conversion of starting material into a higher-running product $(R_f \ 0.56, Et₂O/light$ petroleum 1/1, v/v). The reaction mixture was cooled $(0^{\circ}C)$, *t*-butyl hydroperoxide (0.26) mL) was added and strirring was continued for 30 min after which TLC analysis revealed complete disappearance of the phosphite triester intermediate into a lower-running product $(R_f \ 0.24, Et_2O)$. The reaction mixture was diluted with EtOAc (10 mL), washed with H_2O , dried over $MgSO_4$, filtered and concentrated in vacuo. Compound 37 was obtained as a colorless oil after purification by column chromatography (light petroleum/EtOAc, $3/1\rightarrow 0/1$, v/v) in a 60% yield (86 mg, 68 µmol). ³¹P{¹H} NMR (CDCl₃): δ $-3.2, -3.7, -3.9;$ ES-MS: 1275 $[M+H]^+, 1297$ $[M+$ Na]⁺; ¹³C{¹H} NMR (CDCl₃): δ 137.3 (Cq Bn-2"), 136.4, 135.8, 135.7, 135.4 (6×Cq Bn), 128.7-127.6 (CH Bn), 102.3 (C-1'), 94.5 (C-1"), 80.4, 80.2, 79.1, 78.2, 77.1, 74.9, 72.7, 72.6 (C-2", C-3", C-4", C-5", C-2', C-3', C-4', C-3 propargyl), 74.5, 71.9, 70.6 (C-2 propargyl, $2 \times OCH_2$ $n-Bu$), 69.9, 69.7, 69.6, 69.2, 69.1, 68.9 (7 \times CH₂ Bn), 65.7, 65.4 (C-6", C-5'), 53.1 (C-1 propargyl), 26.9, 25.7 (2 \times CH₂) $n-Bu$). Trisphosphate 37 (86 mg, 68 µmol) was dissolved in a mixture of 1,4-dioxane (5 mL), i-propanol (2.5 mL) and $H₂O$ (1.25 mL) containing NaOAc (66 mg, 0.80 mmol). The mixture was degassed and 10% Pd/C (50 mg), was added and stirred under an atmosphere of H_2 for 16 h. The catalyst was removed by filtration over Glass Fiber (GF/2A, Whatman[®]) and the filtrate was concentrated. Purification was performed as described for the synthesis of cyclophostin 14 to give derivative 15 as a white fluffy foam. Yield 52 mg (57 µmol, 84%, Na⁺-form). ¹H NMR (D₂O, 600 MHz, H-H-COSY): δ 5.15 (s, 1H, H-1'), 5.11 (d, 1H, $H-1''$, $J_{1'',2''}=3.9$ Hz), 4.55 (dd, 1H, H-2', $J_{2',3'}=4.4$ Hz, ${}^{3}J_{P} = 8.5$ Hz), 4.46 (dd, 1H, H-3', $J_{3',4'} = 9.7$ Hz), 4.34 (q, 1H, H-3ⁿ, $J_{2^n,3^n} = J_{3^n,4^n} = {}^3J_p = 9.7$ Hz), 4.27 (m, 1H, H-4¹), 4.02 (t, 1H, $H=5^{n}$, $J_{4^{n}5^{n}}=J_{5^{n}6b^{n}}=9.2$ Hz), 3.91 (d, 1H, H-6a", J=-11.8 Hz), 3.89 (d, 1H, H-5a', $J_{5a',5b'}$ =
-11.6 Hz), 3.80 (q, 1H, H-4^{n 3}J_P=9.6 Hz), 3.69 (dd, 1H, H-2"), $3.67-3.56$ (m, 6H, H-6b", $2 \times OCH_2$ n-Bu, H-1a n -Pr), 3.51 (dd, 1H, H-5b', $J_{4',5b'}$ =4.5 Hz), 3.51 (m, 1H, H-1b n-Pr), $1.79-1.74$; $1.64-1.58$ (2 \times m, 4H, 2 \times CH₂ n-Bu), 1.55 (m, 2H, H-2 n-Pr), 0.86 (t, 3H, H-3 n-Pr, J=7.7 Hz); ¹³C{¹H} NMR (D₂O): δ 106.7 (C-1'), 98.5 (C-1"), 80.5 (C-4'), 77.7, 75.8, 75.2, 71.8 (C-2', C-3', C-2", C-3", C-4", C-5"), 71.0, 70.9, 70.3, 69.9, 69.5 $(2 \times OCH_2 \ n-Bu, C-1 \ n-Pr, C-5', C-6'')$, 26.2, 25.3 (CH₂) $n-Bu$), 22.9 (C-2 n-Pr), 22.9 (C-3 n-Pr); ³¹P NMR (D₂O, 242 MHz, P-H-COSY): δ 2.35 (P-3"), 1.35 (P-2'), 1.11 $(P-4'')$; ES-MS: 647 $[M-H]$ ⁻, 669 $[M-2H+Na]$ ⁻, 691 $[M-3H+2Na]$, 713 $[M-4H+3Na]$. HRMS (ES): Calcd C₁₉H₃₈O₁₉P₃ [M-H]⁻ 663.1219, found 663.1208 (± 0.0023) .

Biological Evaluation

${}^{3}\mathrm{H}\text{-IP}_{3}$ Displacement binding experiments

A 'P₂' fraction of bovine adrenal cortex was prepared as described previously.27 Increasing concentrations of adenophostin A (1) ,²⁸ IP₃ (4), cyclophostin 14, ribophostin $8^{2c,13}$ and cycloribophostin 15 were incubated with a constant amount of 3 H-IP₃ (approx. 9000 d.p.m. per assay; stock:

21 Ci mmol⁻¹; NEN) and adrenal cortex membranes; incubations were stopped after 30 min at 4° C by rapid vacuum filtration.²³ Non-specific binding was defined in the presence of 10 μ M IP₃. The displacement isotherm of each of the ligands (not shown) was used to obtain an estimate of the IC_{50} values (see Table 3) using GraphPad Prism and are given as $-\log$ IC₅₀ values (\pm s.e. mean).

${}^{45}Ca^{2+}$ -Release experiments

Assays were performed using SH-SY5Y human neuroblastoma cells (passage $20-30$) essentially as described previously²⁴ with certain modifications. Confluent monolayers of SH-SY5Y cells were washed and harvested using 10 mM HEPES, 0.9% NaCl, 0.02% EDTA, pH 7.4 and recovered by centrifugation $(400 \times g, 3 \text{ min})$. Cells were resuspended in an `intracellular-like' buffer (ICB: 20 mM HEPES, 135 mM KCl, 2.5 mM $MgCl₂$, 2 mM ATP, 20 μ M CaCl₂, pH 7.1; the free $\lceil Ca^{2+} \rceil$ was buffered to 100–150 nM by addition of EGTA) and centrifuged (400 \times g, 3 min), this latter step was repeated and the final cell pellet was gently resuspended in ICB supplemented with an ATP regenerating system (10 mM phosphocreatine, 10 U mL $^{-1}$ creatine phosphokinase) and permeabilization was achieved by addition of 50 μ g mL⁻¹ α -escin. After 2 min 1 μ Ci mL⁻¹ ${}^{45}Ca^{2+}$ (1000 Ci mmol⁻¹, Amersham, Little Chalfont, UK) was added and the permeabilized cell suspension was added to ICB containing different concentrations of adenophostin A (1), IP₃ (4), cyclophostin 14, ribophostin 8 or cycloribophostin 15. Incubations were continued for 2 min $(\text{IP}_3, 30 \text{ s})$ and samples were then centrifuged $(13,000 \times g, 3 \text{ min})$. A silicone oil mixture $(300 \mu L)$ of Dow-Corning 556/550, 1:1, v/v) was then added to each tube and the samples were recentrifuged $(13,000 \times g, 3 \text{ min})$. The ICB and oil were then aspirated, tubes inverted and allowed to drain for ≥ 60 min before addition of 1.1 mL FloScint IV scintillation cocktail (Packard Bioscience BV, Groningen, the Netherlands). Samples were stored overnight in the dark before scintillation counting. The total releaseable ${}^{45}Ca^{2+}$ pool was defined as that released by addition of $10 \mu M$ ionomycin. Each release isotherm was used to obtain estimates of the EC_{50} value, the slope factor (h) and the maximum obtainable release (expressed as a % of the total ionomycin-releaseable pool) using GraphPad Prism.

Molecular modeling

All calculations were run on IRIS workstations according to Hotoda et al.⁷ Full details on the molecular dynamics simulation will be published elsewhere.²⁹

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